HIGHER ISOPRENOIDS-VIIT

A NEW APPROACH TO THE DEGRADATION OF CYCLOLAUDENOL SIDE-CHAIN‡

CHANDAN SINGH, JOGINDER SINGH and SUKH DEV*
Malti-Chem Research Centre, Nandesari, Vadodara, India

(Received in UK 28 December 1976; Accepted for publication 20 January 1977)

Abstract—An efficient new procedure for the degradation of cyclolaudenyl acetate to 3β - acetoxy - $4.4.14\alpha$ - trimethyl - 9.19 - cyclo - 5α - pregnan - 20 - one and/or 3β - acetoxy - $4.4.14\alpha$ - trimethyl - 9.19 - cyclo - 5α - androstan - 17 - one is described. The key-step is simultaneous base-catelysed dehydrohalogenation and isomerisation of the resulting olefin, of a suitably derived halide.

Side-chain degradation of certain steroids and tetracyclic triterpenes is of considerable interest both from academic and applied points of view. In recent years there has been increased activity in this field and a number of ingenious routes1-3 have been described and these are often more efficient and convenient than the classical Barbier-Wieland degradation or its Meystre-Miescher modification.9 Previously, we described2 a novel approach to such degradations, which had, as its keyfeature, base-catalysed isomerisation of a suitably derived olefin to the C₂₀₍₂₂₎/C₁₇₍₂₀₎ isomer(s). It was found that the most expedient route to the nor-ketones 2 and 3 from cycloartenol 1, lay in the application of this isomerisation-oxidation sequence to cycloartenol itself; since then, we have improved and modified the experimental technique, and these details are reported in the Experimental. However, this isomerisation-oxidation sequence cannot be applied directly to cyclolaudenol 4, another suitable precursor for transformation into 2 or 3, available from opium marc,2 as the isomerisation of the olefinic linkage to the C₂₀₂₂/C₁₂₍₂₀₎ positions is blocked by the thermodynamically stable olefin 5, which must lie on

the base-catalysed isomerisation pathway. Though, we have already described² an indirect application of the isomerisation sequence to cyclolaudenol for the preparation of 2 and 3, we report, in the present work, more direct pathways.

It is obvious that the prime requirement for the application of the isomerisation-oxidation sequence to cyclolaudenol, is the removal of C_{25} , C_{26} and C_{27} in such a way that the resulting tris-nor derivative is an olefin or can be readily converted into one. The successful routes by which this was accomplished are summarised in Fig. 1; the reaction sequence involving intermediate 7 proved most advantageous in terms of yields.

Cyclolaudenol was first converted into the methyl ketone 6 via the Δ²⁴²⁵¹-isomer 5,¹⁰ by a route, already described. This was next reduced with NaBH₄ to furnish alcohol(s) 7, which without further purification was used in the next step. Attempted dehydration of 7 with POCl₃-pyridine (30-100°) gave only the corresponding chloride and practically no olefin was formed. This chloride 8 could be obtained in 75% yield by reacting alcohol 7 with POCl₃-pyridine on a steam-bath for 12 h. It may be pointed out that POCl₃-pyridine¹¹ is a well-established reagent for the dehydration of alcohols and has been successfully employed for the dehydration of tertiary¹²

Part VI: Tetrahedron 33, 0000 (1977).

\$MRC Communication No. 8.

and secondary alcohols,¹³ though with primary alcohols the main product is usually a phosphate ester.¹⁴ In the present instance, the almost exclusive formation of a chloro derivative is rather exceptional,¹³ though chlorides have been known to be formed to a minor extent in such reactions.¹⁶ Dehydrohalogenation of 8 and isomerisation of the resulting olefin to the $\Delta^{20(23)}$ -position was carried out in a single step using N-lithio-ethylenediamine at 120–125° (1 h), when 9 was obtained after acetylation in 90% yield. Ozonolysis of 9 furnished 2, identical in all respects, with the previously reported² preparation.

The above successful dehydrohalogenation-isomerisation prompted us to explore an alternative route, shown in Fig. 1. Cyclolaudenyl acetate 10 on ozonolysis, followed by NaBH₄-reductive work-up.¹⁷ gave 11. β-Fragmentation¹⁸ of this compound with Pb(OAC)₄ and I₂, was investigated in an attempt to obtain 12, which could then be converted into 9. In practice, it was possible to obtain 12 in 40% yield, the remaining product being the expected ether, which is discussed later. As anticipated, 12 was smoothly converted to 9 on exposure to N-lithioethylenediamine, followed by acetylation. Compound 12 was subjected to dehydrohalogenation with 1,5 - diazobicyclo[5.4.0]undec - 5 - ene (DBU)¹⁹ in DMSO to get the olefin 13 in 60% yield, which must have the trans-

and trans - 4 - methyl - 2 - pentene²⁰). This compound was isomerised with N-lithio-ethylenediamine and then acetylated to furnish a mixture of 9 and 14; ozonolysis of this olefin mixture gave the expected 3β - acetoxy - 4.4.14 α - trimethyl - 9.19 - cyclo - 5 α - pregnan - 20 - one 2^2 and 3β - acetoxy - 4.4.14 α - trimethyl - 9.19 - cyclo - 5 α - androstan - 17 - one 3^2 respectively.

As stated earlier, the major product of the action of Pb(OAc)_c- I_2 (h ν) on 11 is an ether, which has the expected structure 15, as revealed by its PMR spectrum. The structure was further confirmed by its acid-catalysed cleavage (p-TsOH, Ac₂O and AcOH)²¹ to 16 (major) and 17, the structures of which are consistent with their spectral and analytical data.

EXPERIMENTAL

All m.ps are uncorrected. Light petroleum refers to the fraction b.p. 60-80°. Optical rotations were measured in CHCl₃ on a Schmidt and Haensch electronic polarimeter model Polatronic 1. TLC was carried out on SiO₂-gel layers (0.25 mm) containing 15% gypsum and activated for 1 h at 100-110°. Silica-gel for column chromatography was activated at 130-140° (6 h) and then standardised. The following instruments were used for spectral data: Perkin-Elmer Infracord model 267 (1R); Perkin-Elmer model R 32 (90 MHz) NMR spectrometer; CEC mass spec-

Fig. 1. Side-chain degradation of cyclolaudenol to 3β - acetoxy - 4,4,14 α - trimethyl - 9,19 - cyclo - 5 α - pregnan - 20 - one.

trometer, model 21-110B (70 eV, direct inlet system). All PMR spectra were recorded using 10-15% soln in CCl₄, unless otherwise stated. While citing PMR data, following abbreviations have been used: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; b, broad. While summarising mass spectral data, besides the molecular ion, ten most abundant ions (above m/e 60) are reported with their relative intensities.

Isomerisation of cycloartenol and subsequent cleavage to 3β - acetoxy - $4.4.14\alpha$ - trimethyl - 9.19 - cyclo - 5α - pregnan - 20 - one 2 and 3β - acetoxy - $4.4.14\alpha$ - trimethyl - 9.19 - cyclo - 5α - androstan - 17 - one 3

Cycloartenol (11.0 g) was added to a stirred soln of N-lithio-ethylenediamine in ethylenediamine (Li 11.0 g, anhydrous ethylenediamine 330 ml; N_3) at 120-125° (bath temp.), and the reaction mixture maintained at this temp. for 20 h. The reaction mixture was cooled to 0°, diluted with ice-water (500 ml) and the product was extracted with isopropyl ether (400 ml × 5). The extract was washed with 10% HCl aq, 5% NaHCO₃ aq, water and brine, dried ($N_{32}SO_4$) and freed of solvent to give a gum (10.94 g) which was directly acetylated (pyridine 26.4 ml, Ac₃O 13.2 ml) to yield 11.7 g of the product.

The above product $(6.50\,\mathrm{g})$ in dry CHCl₁ $(300\,\mathrm{ml})$ was treated with ozonized oxygen $(0.7\,\mathrm{g})$ of O_3/h) at -5° till no more O_3 was absorbed $(1\,\mathrm{h})$. The solvent was flashed off and the crude "ozonide" taken up in acetone $(10\,\mathrm{ml})$, cooled to 0° and treated with Jone's reagent? $(15\,\mathrm{ml})$ till a brown color persisted. The reaction mixture was allowed to stand at room temp. (30°) for 0.5, diluted with H_2O $(100\,\mathrm{ml})$ and the product taken up in isopropyl ether $(100\,\mathrm{ml}\times5)$, which was separated into acidic $(1.36\,\mathrm{g})$ and neutral parts $(4.9\,\mathrm{g})$ with 5% NaOH aq. The neutral material was chromatographed over SiO_2 -gel/IIB $(66.5\times1.5\,\mathrm{cm})$ with TLC monitoring (solvent: 5% EtOAc in C_8H_8 ; two major spots with R_1 (0.27, 0.22):

Fraction 1	light pet.	200 ml × 5	0.149 g, mixture R _f 0.27
Fraction 2	25% C _a H _a in light pet. 50% C _a H _a in	200 ml × 4	1.47 g, essentiall compound with
	light pet.	200 ml × 2	R, 0.27
Fraction 3	50% C ₄ H ₄ in light pet.	200 ml × 4	0.55 g. R _f 0.27
Fraction 4	C.H.	200 ml × 1	0.06 g, mixture
Fraction 5	C.H.	200 ml × 5	1.01 g. R, 0.22
Fraction 6	EtOAc MeOH	200 ml × 1 200 ml × 1	1.39 g, gum, mixture.

 C_{30} -Ketone 2. Fraction 3 was crystallised from ether-MeOH to furnish 2, m.p. 152-153* (lit.2 m.p. 153-154°). Fraction 2 on rechromatogarphy yielded another 1.0 g of 2.

C₁₇-Ketone 3. Recrystallisation of Fraction 5 from aq MeOH yielded pure 3, m.p. 161.5° (lit. m.p. 162-163°).

25,26,27 - Trisnor + 3β + acetoxy + $4,4,14\alpha$ + trimethyl + 9,19 + cyclo + 5α + ergostan + 24 + ol 7

The trisnor-ketone 6² (4.26 g) and NaBH₄ (0.3 g) were reacted in EtOH-dioxane (5:1, 300 ml) at 25° for 4 h and, then worked up in the usual manner to give a solid (3.92 g, m.p. 157-160°), which was recrystallised (ether-MeOH) to furnish 7, m.p. 160-167° (possibly a mixture of C₂₄-epimers). IR (Nujol): OAc 1725, 1240 cm⁻¹; OH 3300, 1020 cm⁻¹. PMR: cyclopropyl CH₂ (1H, d, 0.32 ppm; 14, d, 0.57 ppm; J 4 Hz), tertiary Me's (singlets at 0.83, 0.87, 0.89 and 0.96 ppm), CHMeOH (3H, d, 1.12 ppm; J 6.5 Hz), OAc (3H, s, 2.0 ppm). CHOH (1H, m, 3.64 ppm), CHOAc (1H, m, 4.51 ppm). Mass: mle 458 (M⁺, 4%), 398 (49%), 276 (47%), 175 (56%), 135 (49%), 121 (56%), 109 (55%), 107 (60%), 95 (77%), 69 (100%), 67 (49%). (Found: C, 78.83; H, 11.00, C₁₀H₁₀O₃, requires: C, 78.55; H, 10.99%).

25.26,27 - Trisnor - 3B - acetoxy - 24 - chloro - 4,4,14a - trimethyl - 9,19 - cyclo - 5a - ergostane 8

To a cooled (0°) soln of the above alcohol 7 (10.0 g) in pyridine (50 ml). POCl₃ (15 ml) was introduced (10 min) dropwise with swirling. The reaction mixture was heated on a steam-bath for 12 h, after which it was worked up in the usual manner (extraction with isopropyl ether) to give a solid (7.69, m.p. 160–164°), which was crystallised from ether-MeOH, m.p. 170–172°. IR (Nujol): OAc 1725, 1240 cm⁻¹; C-Cl 760 cm⁻¹. PMR: cyclopropyl CH₂ (1H, d, 0.31 ppm; 1H, d, 0.57 ppm; J 4 Hz), tertiary Me's (singlets at 0.83, 0.86, 0.89 and 0.96 ppm), CHMeCl (3H, d, 1.48 ppm; J 6.5 Hz), OAc (3H, s, 1.97 ppm), CHCl (1H, m, 3.87 ppm), CHOAc (1H, m, 4.47 ppm). Mass: mle 478 (M^{*}, ¹⁵Cl, 4%), 175 (49%), 121 (47%), 109 (51%), 107 (62%), 105 (47%), 95 (100%), 93 (56%), 83 (47%), 81 (56%), 69 (67%), (Found: C, 75.44; H, 10.55; Cl, 7.72. C₁₀H₂₀O₂Cl requires: C, 75.50; H, 10.35; Cl, 7.43%).

26 · Nor · 3β · acetoxy · 4,4,14α · trimethyl · 9,19 · cyclo · 5α · ergostan · 25 · ol 11

Cyclolaudenyl acetate 10 (5.0 g, m.p. 120-121°) in CHCl₃ (200 ml) was ozonised at -5° in the usual fashion. The soln was concentrated to 50 ml by solvent removal under vacuum at room temp., and then cooled (0°) and, a soln of NaBH₄ (4.0 g in 75 ml of 85% aq EtOH) added (N₅) under stirring. After stirring for 5 h at 30°, the reaction mixture was worked up in the usual manner

to give semi-solid product (5.04 g), which was chromatographed on SiO₂-gel/IIB (65 × 1.5 cm) using light petroleum, light petroleum with increasing amounts of C_aH_a and pure C_aH_a as eluants. C_aH_a cuts (200 ml × 8) gave the required product (3.15 g, m.p. 143–145°, epimeric pair), which was crystallised from light petroleum, colorless needles, m.p. 147.5–150°, $\{a\}_D$ +57.0°. (c. 0.8%). IR (Nujol): OAc 1730, 1240 cm⁻¹; OH 3350, 1095 cm⁻¹. PMR: cyclopropyl CH₂ (1H, d, 0.32 ppm; 1H, d, 0.59 ppm; J 4Hz), tertiary Me's (singlet at 0.86, 0.88, 0.91 and 0.97 ppm), CHMeOH (C_{35} -epimers; 3H, two doublets centred at 1.05 and 1.12 ppm, J 7 Hz each), OAc (3H, s, 2.0 ppm), CHOM (1H, m, 4.50 ppm), Mass: m/e 486 (M°, 8%), 426 (87%), 304 (56%), 175 (58%), 121 (48%), 119 (45%), 109 (65%), 107 (64%), 95 (100%), 81 (53%), 69 (80%). (Found: C, 78.86; H, 10.87, $C_{12}H_{34}O_3$ requires: C, 78.96; H, 11.18%).

Action of lead tetraacetate-iodine on 11

A mixture of the above alcohol 11 (0.972 g, 0.002 mole), Pb (OAc), (0.986 g, 0.0022 mole) and jodine (0.289 g, 0.0022 g atom) in dry cyclohexane (100 ml) was stirred (N2) and exposed to a 250 Watt tungsten lamp from below, when refluxing ensued. After stirring and refluxing for a total of 3 h, ethylene glycol (5.0 ml) was added to destroy any excess Pb(OAc)4. The reaction mixture was diluted with water (100 ml), the upper layer separated and the aq phase extracted with isopropyl ether (50 ml × 4). The combined organic phases were washed with 10% Na₂S₂O₃ aq. water, brine and dried (Na₂SO₄). The solvent was flashed off and the residue (1.47 g) in isopropyl ether (10.0 ml) filtered through a bed (10.0 cm × 1.5 cm) of Al₂O₃/III, which was further washed with the same solvent (200 ml). The ether soln was freed of the solvent and the residue (1.4 g) chromatographed over SiO₂ gel/IIB (30.0×2.5 cm), while monitoring with TLC (solvent: 5% EtOAc in C.H.).

Fraction 1	light pet. 25% C.H. in	200 ml	
	light pet.	200 ml × 2	0.1 g, mixture
Fraction 2	25% C.H. in)	
	light pet.	200 m!	0.349 g, pure,
	50% CaHa in	ĺ	• •
	light pet.	200 ml	R, 0.66
Fraction 3	50% C.H. in		.,
	light pet.	200 ml	0.250 g, mixture
Fraction 4	C.H.	200 ml × 3	0.337 g, pure,
			R, 0.48
Fraction 5	C ₄ H ₄	200 ml × 3	0.05 g, mixture
Fraction 6	C ₄ H ₄	200 ml	0.16 g, R, 0.22
	•		unchanged 11

25,26,27 · Trisnor · 3β · acetoxy · 24 · iodo · 4,4,14a · trimethyl · 9,19 · cyclo · 5a · ergostane 12

Recrystallisation of Fraction 2 from isopropyl ether yielded 12, m.p. $114-116^\circ$ (dec.), $[\alpha]_{\rm b}$ +33.55° (c, 0.77%). IR (Nujol): OAc 1730, 1250 cm⁻¹. PMR: cyclopropyl CH₂ (1H, d, 0.32 ppm; 1H, d, 0.57 ppm; J 4 Hz), tertiary Me's (singlets at 0.85, 0.90, 0.92 and 0.97 ppm). CHMeI (3H, d, 1.93 ppm; J 6.5 Hz), OAc (3H, s, 1.99 ppm). CHI (1H, m, 4.1 ppm), CHOAc (1H, m, 4.5 ppm). Mass: m/e 568 (M°, 15%), 508 (66%), 175 (63%), 121 (63%), 109 (73%), 107 (82%), 105 (54%), 95 (100%), 93 (63%), 11 (70%), 69 (79%). (Found: C, 63.24; H, 8.95; I, 20.88. $C_{10}H_{20}O_{2}I$ requires: C, 63.36; H, 8.68; I, 22.31%).

26 - Nor - 3β - acetoxy - 22,25 - epoxy - 4,4,14α - trimethyl - 9,19 - cyclo - 5α - ergostane 15

Fraction 4 on crystallisation from isopropyl ether-MeOH gave the ether 15, m.p. 162.5– 164° , $\{\alpha\}_D + 35.11^\circ$ (c, 0.65%). IR (Nujol): OAc 1725, 1240 cm⁻¹; C-O-C 1080, 1100 cm⁻¹. PMR: cyclopropyl CH₂ (1H, d, 0.32 ppm; 1H, d, 0.58 ppm; J 4 Hz), tertiary Me's (3H, singlets at 0.85, 1.0 ppm; 6H, s, 0.90 ppm), MeHCOC (3H, d, 1.16 ppm; J 6.5 Hz), OAc (3H, s, 1.99 ppm), HC₂₇–O-C (1H, m, 3.25 ppm), HC₂₇–O-C (1H, m, 3.85 ppm). CHOAc (1H, m, 4.48 ppm). Mass: mle 484 (M*, 5%), 469 (1.5%), 424 (11%), 302 (5%), 203 (4%). 175 (5%), 147 (5%), 126 (22%), 99 (100%). 81

(12%), 69 (8%). (Found: C, 79.65; H, 10.09. C₃₂H₃₂O₃ requires: C, 79.28; H, 10.81%).

25,26,27 - Trisnor - 3β - acetoxy - 4,4,14α - trimethyl - 9,19 - cyclo - 5α - ergost - 20(22) - ene 9

From C₃₄-chloride 8. To a soln of N-lithio-ethylenediamine in ethylenediamine (Li, 3 g; NH₂CH₂CH₂NH₂, 90 ml), chloride 8 (4.9 g) was added with stirring at 125° (bath temp). The reaction mixture was heated at that temp for 1 h and then worked up in the usual manner to give a gum, which was acetylated (Ac₂O, pyridine) to furnish a product (4.3 g), which soon crystallised. Recrystallisation from ether-MeOH gave pure 9, m.p. 102-105°. IR (Nujol): OAc 1725, 1240 cm⁻¹. PMR: cyclopropyl CH₂ (1H, d, 0.33 ppm; 1H, d, 0.59 ppm; J 4 Hz), Me's (singlets at 0.83, 0.87, 0.89, 0.96 and 0.98 ppm), Me-C=C (bs, 1.64 ppm), OAc (3H, s, 1.98 ppm), CHOAc (1H, m, 4.48 ppm), C=CH (1H, m, 5.22 ppm). Mass: mle 440 (M⁻, 30%), 380 (64%), 175 (68%), 121 (83%), 107 (84%), 105 (65%), 93 (94%), 91 (71%), 83 (100%), 81 (80%), 69 (65%). (Found: C, 82.26; H, 11.10. C₁₀H₄₄O₂ requires: C, 81.76; H, 10.98%).

From C₂₄-iodide 12. C₂₄-iodide 12 (0.3 g) was likewise treated (Li, 1.0 g; NH₂CH₂CH₂NH₂, 45 ml) as above to give, after acetylation, 0.213 g of crude 9, which was purified as above.

25,26,27 - Trisnor - 3β - acetoxy - 4,4,14α - trimethyl - 9,19 - cyclo - 5α - ergost - 23 - ene 13

C_{3c}-iodide 12 (0.5 g) in DMSO (5.0 ml) containing DBU (0.17 g) was stirred and heated at 80-90° for 13 h (N₂) and then worked up in the usual manner to give a product, which was chromatographed on Al₂O₃/III (20.0 × 1.5 cm). Light petroleum (100 ml) eluted a product (TLC pure; solvent, 5% EtOAc in C_6H_6 ; R_c 0.66; 0.235 g) which was crystallised from isopropyl ether-MeOH: m.p. 119.5-120.5°. PMR: Cyclopropyl CH₂ (1H, d. 0.33 ppm; 1H, d. 0.53 ppm; J 4 Hz), tertiary Me's (12H, bs. 0.86 ppm), Me-CH=CH (bd. 1.65 ppm; J 7 Hz), OAc (3H, s. 1.97 ppm), CHOAc (1H, m. 4.43 ppm), -CH=CH- (2H, m. 5.35 ppm), Mass: mie 440 (M°, 38%), 425 (45%), 380 (14%), 365 (34%), 297 (5%), 215 (27%), 175 (38%), 121 (65%), 95 (100%), 81 (79%), 69 (93%). (Found: C, 81.85; H, 10.76. $C_{10}H_{10}O_{2}$ requires: C, 81.76; H, 10.98%).

 3β - Acetoxy - 4,4,14 α - trimethyl - 9,19 - cyclo - 5α - pregnan - 20 - one 2

A soln of olefin 9 (4.3 g) in CHCl₃ (200 ml) was treated with ozonised O₂ (0.48 g O₃/h) at 10° till O₃ passed freely. This was worked up and oxidised by Jones reagent, as detailed earlier, to give a neutral fraction (gum, 3.0 g), which after chromatographic purification (see above) furnished pure 2 (2.0 g), m.p. 153-154° (lit.² m.p. 153-154°).

 3β - Acetoxy - 4,4,14 α - trimethyl - 9,19 - cyclo + 5α - androstan - 17 - one 3

Olefin 13 (0.15 g) was isomerised (Li, 1.0 g; NH₂CH₂CH₂NH₂, 40 ml) at 120-125° (bath) for 25 h and the product, acetylated, ozonolyzed and the product (neutral, 0.12 g) separated by chromatography exactly as described earlier to give 2 (0.035 mg, m.p. 152-153°) and 3 (0.048 g, m.p. 160-163°).

Cleavage of ether 15

Ether 15 (0.5 g) in gl. AcOH (15.0 ml) containing Ac₃O (7.5 ml) and p-TsOH (0.049 g) was heated (60-65°) with stirring for 43 h (N₃). On usual work-up a semi-solid (0.52 g), showing on TLC (solvent: 5% EtOAc in C₄H₂) essentially two spots (R_r , 0.60 major; R_r , 0.38, minor), was obtained. This was chromatographed on SiO₂-gel/IIB (43.0 × 2.0 cm) using light petroleum, light petroleum containing increasing quantities of C₄H₄, C₄H₄ and C₄H₄ containing increasing amounts of EtOAc, as eluant.

26 - Nor - 38,22 - diacetoxy - 4,4,14\alpha - trimethyl - 9,19 - cyclo - 5 - ergost - 24 - ene 16

The material eluted with 50% C_aH_a in light petroleum (200 ml × 2; 0.331 g, m.p. 175-178°) on crystallisation gave pure 16, m.p. 184-186°, $\{\alpha\}_D \neq 26.06^\circ$ (c, 0.53%). IR (Nujol): OAc 1735, 1245 cm ¹. PMR: cyclopropyl CH₂ (1H. d, 0.33 ppm; 1H, d,

0.61 ppm; J 4 Hz), tertiary Me's (singlets at 0.86, 0.87, 0.90 and 0.99 ppm). Me-CH=C (d, 1.75 ppm; J 7 Hz). OAc (3H, s, 1.93 ppm; 3H, s, 2.0 ppm), HC₂-OAc (1H, m, 4.48 ppm), HC₂-OAc (1H, m, 5.14 ppm), C=CH (1H, m, 5.22 ppm). (Found: C, 77.47; H, 10.04. $C_{24}H_{34}O_{4}$ requires: C, 77.60; H, 10.27%).

5% EtOAc in C.H. (200 ml) eluted 0.12 g of a material (R₁, 0.38) which, though essentially pure (TLC), could not be induced to crystallise. From its PMR data (cyclopropyl CH₂, 1H doublets at 0.33, 0.61 ppm; four tertiary Me's, 0.86, 0.90, 0.90, 0.97 ppm; MeCHOAc, d, 1.13 ppm, J 6.5 Hz; Me-C=C, 1.7 ppm; two OAc, singlets at 2.04, 2.06 ppm; HC₂-OAc, m, 4.50 ppm; HC₂-OAc, m, 5.0 ppm; C=CH, m, 5.10 ppm) it is clearly 17.

REFERENCES

- ¹For a recent survey see: A. S. Narula, J. Sci. Indust. Res. 31, 423 (1972).
- ²A. S. Narula and Sukh Dev, Tetrahedron 27, 1119 (1971).
- ³L. H. Briggs, J. P. Bartley and P. S. Rutledge, J. Chem. Soc. Perkin I 806 (1973).
- ⁴M. Fetizon, F. J. Kakis and V. Ignatiadou-Ragoussis, J. Org. Chem. 38, 1732, 4308 (1973); 39, 1959 (1974).
- ⁵M. Fetizon, F. J. Kakis and V. Ignatiadou-Ragoussis, Tetrahedron 30, 3981 (1974).
- *B. Ganem and M. S. Kellog, J. Org. Chem. 39, 575 (1974),
- ⁷J. Bernassau and M. Fetizon, Synthesis 795 (1975).
- *P. Barbier and R. Locquin, C.R. Acad. Sci. 156, 1443 (1913); H. Wieland, O. Schlichting and R. Jacobi, Z. Physiol. 161, 80 (1926).
- *Ch. Meystre, L. Ehmann, R. Neher and K. Miescher, Helv. Chim. Acta 28, 1252 (1945).

- ¹⁰This compound is known to occur in nature and has been named cyclobranol: T. Endo, Y. Inada and S. Naito, Chem. Abstr. 73, 45627 (1970).
- ¹¹A. Butenandt and J. Schmidt-Thome, Ber. Disch. Chem. Ges. 72, 182 (1939).
- ¹²D. H. R. Barton, A. Das, Campos-Neves and R. C. Cookson, J. Chem. Soc. 3500 (1956).
- ³³A. Crawshaw, H. B. Henbest and E. R. H. Jones, J. Chem. Soc. 731 (1954).
- ¹⁴C. H. G. Hands, Chem. and Ind. 1032 (1972).
- ¹⁶cf. B. Fuchs and H. J. E. Loewenthal, Tetrahedron 11, 199 (1960).
- ¹⁴See e.g.: K. Heusler and A. Wettstein, Helv. Chim. Acta 35, 284 (1952); E. Klinotova, N. Hovorkova, J. Klinot and A. Vystrcil, Coll. Czech. Chem. Comm. 38, 1179 (1973).
- ¹⁷John A. Sousa and Aaron L. Bluhm, J. Org. Chem. 25, 108 (1960).
- ¹⁸For a review see: K. Heusler and J. Kalvoda, Angew. Chem. Intern. Edit. 3, 525 (1964); M. I.j. Mihailovic and Z. Cekovic, Synthesis 209 (1970).
- ¹⁹H. Oediger and Fr. Moller, Angew. Chem. Intern. Edit. 6, 76 (1967).
- ²⁰C. J. Pouchert and J. R. Campbell, *The Aldrich Library of NMR Spectra*, Vol. 1, p. 24. Aldrich Chemical Company, Milwaukee (1974).
- ²¹J. F. Bagli, P. F. Morand and R. Gaudey, J. Org. Chem. 28, 1207 (1963).
- ²²R. Hernandez, R. Hernandez, Jr. and L. R. Axelrod, *Analyt. Chem.* 33, 370 (1961).
- ²¹A. S. Narula and Sukh Dev, Tetrahedron Letters 1733 (1969).